

Selected Thermodynamic Data p1

| Substance & State | ΔH (kJ/mol) | ΔG (kJ/mol) | S (J/molK) | Substance & State | ΔH (kJ/mol) | ΔG (kJ/mol) | S (J/molK) |
|---|------------------------|------------------------|---------------|---|------------------------|------------------------|---------------|
| Aluminum | | | | Calcium (cont) | | | |
| Al(s) | 0 | 0 | 28 | CaSiO ₃ (s) | -1630 | -1550 | 84 |
| Al ³⁺ (aq) | -531 | -485 | -322 | CaCl ₂ (s) | -796 | -748 | 105 |
| Al ₂ O ₃ (s) | -1676 | -1582 | 51 | Carbon | | | |
| Al(OH) ₃ (s) | -1277 | | | C(s)(graphite) | 0 | 0 | 6 |
| AlCl ₃ (s) | -704 | -629 | 111 | C(s)(diamond) | 2 | 3 | 2 |
| Barium | | | | CO(g) | -110.5 | -137 | 198 |
| Ba(s) | 0 | 0 | 67 | CO ₂ (g) | -393.5 | -394 | 214 |
| Ba ⁺² (aq) | -538 | -561 | 10 | CO ₃ ⁻² (aq) | -677 | -528 | 57 |
| BaCO ₃ (s) | -1219 | -1139 | 112 | CH ₄ (g) | -75 | -51 | 186 |
| BaO(s) | -582 | -552 | 70 | CH ₃ OH(g) | -201 | -163 | 240 |
| Ba(OH) ₂ (s) | -946 | | | CH ₃ OH(l) | -239 | -166 | 127 |
| BaSO ₄ (s) | -1465 | -1353 | 132 | H ₂ CO(g) | -116 | -110 | 219 |
| BaCl ₂ | -859 | -810 | 124 | HCOOH(g) | -363 | -351 | 249 |
| Beryllium | | | | HCN(g) | 135.1 | 125 | 202 |
| Be(s) | 0 | 0 | 10 | C ₂ H ₂ (g) | 227 | 209 | 201 |
| BeO(s) | -599 | -569 | 14 | C ₂ H ₄ (g) | 52 | 68 | 219 |
| Be(OH) ₂ (s) | -904 | -815 | 47 | CH ₃ CHO(g) | -166 | -129 | 250 |
| Bromine | | | | C ₂ H ₅ OH(l) | -278 | -175 | 161 |
| Br ₂ (l) | 0 | 0 | 152 | C ₂ H ₆ (g) | -84.7 | -32.9 | 229.5 |
| Br ₂ (g) | 31 | 3 | 245 | C ₃ H ₆ (g) | 20.9 | 62.7 | 266.9 |
| Br ₂ (aq) | -3 | 4 | 130 | C ₃ H ₈ (g) | -104 | -24 | 270 |
| Br ⁻ (aq) | -121 | -104 | 82 | C ₂ H ₄ O(g) ethylene oxide | -53 | -13 | 242 |
| HBr (g) | -36 | -53 | 199 | CH ₂ =CHCN(g) | 185 | 195.4 | 274 |
| Cadmium | | | | CH ₃ COOH(aq) | -484 | -389 | 160 |
| Cd(s) | 0 | 0 | 52 | C ₆ H ₁₂ O ₆ (s) | -1275 | -911 | 212 |
| Cd ⁺² (aq) | -76 | -78 | -73 | CCl ₄ (l) | -135 | -65 | 216 |
| CdO(s) | -258 | -228 | 55 | CHCl ₃ (l) | -135 | -65 | 202 |
| Cd(OH) ₂ (s) | -561 | -474 | 96 | Chlorine | | | |
| CdS(s) | -162 | -156 | 65 | Cl ₂ (g) | 0 | 0 | 223 |
| CdSO ₄ (s) | -935 | -823 | 123 | Cl ₂ (aq) | -23 | 7 | 121 |
| CdCl ₂ (s) | -392 | -344 | 115 | Cl ⁻ (aq) | -167 | -131 | 57 |
| Calcium | | | | HCl(g) | -92 | -95 | 187 |
| Ca(s) | 0 | 0 | 41 | ClO ₃ ⁻ (aq) | -104 | -8 | 162 |
| Ca ⁺² (aq) | -543 | -554 | -53 | ClO ₄ ⁻ (aq) | -129 | -8.5 | 182 |
| CaC ₂ (s) | -63 | -68 | 70 | Chromium | | | |
| CaCO ₃ (s) | -1207 | -1129 | 93 | Cr(s) | 0 | 0 | 24 |
| CaO(s) | -635 | -604 | 40 | Cr ₂ O ₃ (s) | -1128 | -1047 | 81 |
| Ca(OH) ₂ (s) | -987 | -899 | 83 | CrO ₃ (s) | -579 | -502 | 72 |
| Ca ₃ (PO ₄) ₂ (s) | -4126 | -3890 | 241 | CrO ₄ ⁻ (aq) | -881 | -728 | 50 |
| CaSO ₄ (s) | -1433 | -1320 | 107 | Cr ₂ O ₇ ⁻² (aq) | -1490 | -1301 | 262 |

Selected Thermodynamic Data p2

| Substance & State | ΔH (kJ/mol) | ΔG (kJ/mol) | S (J/molK) | Substance & State | ΔH (kJ/mol) | ΔG (kJ/mol) | S (J/molK) |
|---|------------------------|------------------------|---------------|-------------------------------------|------------------------|------------------------|---------------|
| Copper | | | | Lead | | | |
| Cu(s) | 0 | 0 | 33 | Pb(s) | 0 | 0 | 65 |
| Cu ⁺ (aq) | 72 | 50 | 41 | Pb ²⁺ (aq) | -1.7 | -24 | 10.5 |
| Cu ²⁺ (aq) | 65 | 65.5 | -100 | PbO ₂ (s) | -277 | -217 | 69 |
| CuCO ₃ (s) | -595 | -518 | 88 | PbS(s) | -100 | -99 | 91 |
| Cu ₂ O(s) | -170 | -148 | 93 | PbSO ₄ (s) | -920 | -813 | 149 |
| CuO(s) | -156 | -128 | 43 | PbBr ₂ (s) | -279 | -262 | 161.5 |
| Cu(OH) ₂ (s) | -450 | -372 | 108 | PbCl ₂ (s) | -359 | -314 | 136 |
| CuS(s) | -49 | -49 | 67 | | | | |
| CuSO ₄ (s) | -771 | -662 | 108 | Magnesium | | | |
| | | | | Mg(s) | 0 | 0 | 33 |
| Fluorine | | | | Mg ²⁺ (aq) | -467 | -455 | -138 |
| F ₂ (g) | 0 | 0 | 203 | MgCO ₃ (s) | -1113 | -1029 | 66 |
| F(aq) | -333 | -279 | -14 | MgO(s) | -602 | -569 | 27 |
| HF(g) | -271 | -273 | 174 | Mg(OH) ₂ (s) | -925 | -834 | 64 |
| | | | | MgCl ₂ (s) | -641 | -592 | 90 |
| Hydrogen | | | | MgSO ₄ (s) | -1285 | -1171 | 92 |
| H ₂ (g) | 0 | 0 | 131 | | | | |
| H(g) | 217 | 203 | 115 | Manganese | | | |
| H ⁺ (aq) | 0 | 0 | 0 | Mn(s) | 0 | 0 | 32 |
| OH ⁻ (aq) | -230 | -157 | -11 | MnO(s) | -385 | -363 | 60 |
| H ₂ O(l) | -286 | -237 | 70 | Mn ₃ O ₄ (s) | -1387 | -1280 | 149 |
| H ₂ O(g) | -242 | -229 | 189 | Mn ₂ O ₃ (s) | -971 | -893 | 110 |
| HCO ₃ ⁻ (aq) | -692 | -587 | 91 | MnO ₂ (s) | -521 | -466 | 53 |
| H ₂ O ₂ (l) | -188 | -1130 | 90 | MnO ₄ ⁻ (aq) | -543 | -449 | 190 |
| | | | | Mn ²⁺ (aq) | -221 | -228 | -74 |
| Iodine | | | | | | | |
| I ₂ (s) | 0 | 0 | 116 | Mercury | | | |
| I ₂ (g) | 62 | 19 | 261 | Hg(l) | 0 | 0 | 76 |
| I ₂ (aq) | 23 | 16 | 137 | Hg ₂ Cl ₂ (s) | -265 | -211 | 196 |
| I ⁻ (aq) | -55 | -52 | 106 | HgCl ₂ (s) | -230 | -184 | 144 |
| HI(g) | 26.5 | 1.7 | 206.5 | HgO(s) | -90 | -59 | 70 |
| | | | | HgS(s) | -58 | -49 | 78 |
| Iron | | | | Hg ²⁺ (aq) | 171 | 164 | -32 |
| Fe(s) | 0 | 0 | 27 | | | | |
| Fe ²⁺ (aq) | -89 | -79 | -138 | Nickel | | | |
| Fe ³⁺ (aq) | -48.5 | -5 | -316 | Ni(s) | 0 | 0 | 30 |
| Fe(OH) ₃ (s) | -823 | -697 | 107 | NiCl ₂ (s) | -316 | -272 | 107 |
| Fe ₃ C(s) | 21 | 15 | 108 | NiO(s) | -241 | -213 | 38 |
| Fe _{0.95} O(s)(wustite) | -264 | -240 | 59 | Ni(OH) ₂ (s) | -538 | -453 | 79 |
| FeO(s) | -272 | -255 | 61 | NiS(s) | -93 | -90 | 53 |
| Fe ₃ O ₄ (s) (magnetic) | -1117 | -1013 | 146 | Ni ²⁺ (aq) | -54 | -46 | -129 |
| Fe ₂ O ₃ (s) (hematite) | -826 | -740 | 90 | | | | |
| FeS(s) | -95 | -97 | 67 | | | | |
| FeS ₂ (s) | -178 | -166 | 53 | | | | |
| FeSO ₄ (s) | -929 | -825 | 121 | | | | |

Selected Thermodynamic Data p3

| Substance & State | ΔH (kJ/mol) | ΔG (kJ/mol) | S (J/molK) | Substance & State | ΔH (kJ/mol) | ΔG (kJ/mol) | S (J/molK) |
|---|------------------------|------------------------|---------------|--------------------------------------|------------------------|------------------------|---------------|
| Nitrogen | | | | Potassium | | | |
| N ₂ (g) | 0 | 0 | 192 | K(s) | 0 | 0 | 64 |
| NH ₃ (g) | -46 | -17 | 193 | K ⁺ (aq) | -252 | -283 | 102.5 |
| NH ₃ (aq) | -80 | -27 | 111 | KBr(s) | -394 | -381 | 96 |
| NH ⁴⁺ (aq) | -132 | -79 | 113 | KCl(s) | -436 | -408 | 83 |
| NO(g) | 90 | 87 | 211 | KClO ₃ (s) | -391 | -290 | 143 |
| NO ₂ (g) | 34 | 52 | 240 | KClO ₄ (s) | -433 | -304 | 151 |
| N ₂ O(g) | 82 | 104 | 220 | KNO ₃ (s) | -370 | -395 | 133 |
| N ₂ O ₄ (g) | 10 | 98 | 304 | K ₂ O(s) | -361 | -322 | 98 |
| N ₂ O ₄ (l) | -20 | 97 | 209 | K ₂ O ₂ (s) | -496 | -430 | 113 |
| N ₂ O ₅ (s) | -42 | 134 | 178 | KO ₂ (s) | -283 | -238 | 117 |
| N ₂ H ₄ (l) | 51 | 149 | 121 | KOH(s) | -425 | -379 | 79 |
| N ₂ H ₃ CH ₃ (l) | 54 | 180 | 166 | KOH(aq) | -481 | -440 | 9.2 |
| HNO ₃ (aq) | -207 | -111 | 146 | Silicon | | | |
| HNO ₃ (l) | -174 | -81 | 156 | Si(s) | 0 | 0 | 19 |
| NH ₄ ClO ₄ (s) | -295 | -89 | 186 | SiO ₂ (s) (quartz) | -911 | -856 | 42 |
| NH ₄ Cl(s) | -314 | -203 | 96 | SiCl ₄ (l) | -687 | -620 | 240 |
| NH ₄ NO ₃ (s) | -366 | -184 | 151 | Silver | | | |
| NO ²⁻ (aq) | -105 | -32 | 123 | Ag(s) | 0 | 0 | 43 |
| NO ³⁻ (aq) | -205 | -109 | 146 | Ag ⁺ (aq) | 105 | 77 | 73 |
| Oxygen | | | | AgBr(s) | -100 | -97 | 107 |
| O ₂ (g) | 0 | 0 | 205 | AgCN(s) | 146 | 164 | 84 |
| O(g) | 249 | 232 | 161 | AgCl(s) | -127 | -110 | 96 |
| O ₃ (g) | 143 | 163 | 239 | Ag ₂ CrO ₄ (s) | -712 | -622 | 217 |
| OH ⁻ (aq) | -230 | -157 | -11 | AgI(s) | -62 | -66 | 115 |
| Phosphorus | | | | Ag ₂ O(s) | -31 | -11 | 122 |
| P(s) (white) | 0 | 0 | 41 | Ag ₂ S(s) | -32 | -40 | 146 |
| P(s) (red) | -18 | -12 | 23 | AgNO ₃ (s) | -124 | -33 | 141 |
| P(s) (black) | -39 | -33 | 23 | Sodium | | | |
| P ₄ (g) | 59 | 24 | 280 | Na(s) | 0 | 0 | 51 |
| PF ₅ (g) | -1578 | -1509 | 296 | Na ⁺ (aq) | -240 | -262 | 59 |
| PH ₃ (g) | 5 | 13 | 210 | NaBr(s) | -360 | -347 | 84 |
| H ₃ PO ₄ (s) | -1279 | -1119 | 110 | Na ₂ CO ₃ (s) | -1131 | -1048 | 136 |
| H ₃ PO ₄ (l) | -1267 | | | NaHCO ₃ (s) | -948 | -852 | 102 |
| H ₃ PO ₄ (aq) | -1288 | -1143 | 158 | NaCl(s) | -411 | -384 | 72 |
| P ₄ O ₁₀ (s) | -2984 | -2698 | 229 | NaF | -574 | -544 | 51.5 |
| H ₂ PO ₄ ⁻ (aq) | -1296 | -1130 | 90 | NaH(s) | -56 | -33 | 40 |
| HPO ₄ ²⁻ (aq) | -1292 | -1089 | -33.5 | NaI(s) | -288 | -282 | 91 |
| PCl ₃ (g) | -287 | -268 | 312 | NaNO ₂ (s) | -359 | | |
| PCl ₅ (g) | -375 | -305 | 365 | NaNO ₃ (s) | -467 | -366 | 116 |
| PO ₄ ³⁻ (aq) | -1277 | -1019 | -222 | Na ₂ O(s) | -416 | -377 | 73 |
| | | | | Na ₂ O ₂ (s) | -515 | -451 | 95 |
| | | | | NaOH(s) | -427 | -381 | 64 |

Selected Thermodynamic Data p4

| Substance & State | ΔH (kJ/mol) | ΔG (kJ/mol) | S (J/molK) | Substance & State | ΔH (kJ/mol) | ΔG (kJ/mol) | S (J/molK) |
|-------------------------------------|------------------------|------------------------|---------------|-------------------------|------------------------|------------------------|---------------|
| Sodium (cont) | | | | Zinc | | | |
| NaOH(aq) | -470 | -419 | 50 | Zn(s) | 0 | 0 | 42 |
| Sulfur | | | | Zn ²⁺ (aq) | -154 | -147 | -112 |
| S(s) (rhombic) | 0 | 0 | 32 | ZnO(s) | -348 | -318 | 44 |
| S(s) (monoclinic) | 0.3 | 0.1 | 33 | Zn(OH) ₂ (s) | -642 | | |
| S ²⁻ (aq) | 33 | 86 | -15 | ZnI ₂ (s) | -208 | -209 | 161 |
| S ₈ (g) | 102 | 50 | 431 | ZnS(s) (wurtzite) | -193 | | |
| SF ₆ (g) | -1209 | -1105 | 292 | ZnS(s) (zinc blende) | -206 | -201 | 58 |
| H ₂ S(g) | -21 | -34 | 206 | ZnSO ₄ (s) | -983 | -874 | 120 |
| SO ₂ (g) | -297 | -300 | 248 | | | | |
| SO ₃ (g) | -396 | -371 | 257 | | | | |
| SO ₄ ²⁻ (aq) | -909 | -745 | 20 | | | | |
| HSO ₄ ⁻ (aq) | -887 | -756 | 132 | | | | |
| H ₂ SO ₄ (l) | -814 | -690 | 157 | | | | |
| H ₂ SO ₄ (aq) | -909 | -745 | 20 | | | | |
| Tin | | | | | | | |
| Sn(s) (white) | 0 | 0 | 52 | | | | |
| Sn(s) (gray) | -2 | 0.1 | 44 | | | | |
| Sn ²⁺ (aq) | -9 | -27 | -17 | | | | |
| SnO(s) | -285 | -257 | 56 | | | | |
| SnO ₂ (s) | -581 | -520 | 52 | | | | |
| Sn(OH) ₂ (s) | -561 | -492 | 155 | | | | |
| Titanium | | | | | | | |
| TiCl ₄ (g) | -763 | -727 | 355 | | | | |
| TiO ₂ (s) | -945 | -890 | 50 | | | | |
| Uranium | | | | | | | |
| U(s) | 0 | 0 | 50 | | | | |
| UF ₆ (s) | -2137 | -2008 | 228 | | | | |
| UF ₆ (g) | -2113 | -2029 | 380 | | | | |
| UO ₂ (s) | -1084 | -1029 | 78 | | | | |
| U ₃ O ₈ (s) | -3575 | -3393 | 282 | | | | |
| UO ₃ (s) | -1230 | -1150 | 99 | | | | |
| Xenon | | | | | | | |
| Xe(g) | 0 | 0 | 170 | | | | |
| XeF ₂ (g) | -108 | -48 | 254 | | | | |
| XeF ₄ (s) | -251 | -121 | 146 | | | | |
| XeF ₆ (g) | -294 | | | | | | |
| XeO ₃ (s) | 402 | | | | | | |