



$$D = \frac{M}{V}$$

$$V = \frac{M}{D}$$

$$M = D \times V$$

①  $D = ?$   
 $M = 43.2g$   
 $V = 96.5mL$

$$D = \frac{M}{V} = \frac{43.2g}{96.5mL} = \boxed{0.4489/mL}$$

②  $D = ?$   
 $M = 4.922g$   
 $V = 4.0L$

$$D = \frac{M}{V} = \frac{4.922g}{4.0L} = \boxed{1.2319/L}$$

-or-  $D = \frac{M}{V} = \frac{4.922g}{4000mL} = \boxed{.0012319/mL}$

③  $D = 13.6g/mL$   
 $M = 2242g$   
 $V = ?$

$$V = \frac{M}{D} = \frac{2242g}{13.6g/mL} = \boxed{164.9mL}$$

④  $D = .880g/cm^3$   
 $M = 54g$   
 $V = ?$

$$V = \frac{M}{D} = \frac{54g}{.880g/cm^3} = \boxed{61.4cm^3}$$

⑤  $D = 1.25g/mL$   
 $M = ?$   
 $V = 84mL$

$$M = D \times V = (1.25g/mL) \times 84mL = \boxed{105g}$$

⑥  $D = 1.43g/L$   
 $M = ?$   
 $V = 25mL = .025L$

$$M = D \times V = (1.43g/L) \times (.025L) = \boxed{.0358g}$$

\* I am changing the volume from mL to L so the units of volume (L) agree with the density units (g/L).

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	<u>sample A</u>		<u>sample B</u>		<u>sample C</u>
	$D = \frac{M}{V} = \frac{85g}{116ml} = \underline{.733g/ml}$		$D = \frac{M}{V} = \frac{101g}{168ml} = \underline{.602g/ml}$		$D = \frac{M}{V} = \frac{115g}{158ml}$
					$D = \underline{.728g/ml}$
	$\frac{V}{M}$				
A	116ml	85g			
B	168ml	101g			
C	158ml	115g			

Samples A and C could be the same because their densities are very close.

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$D = 19.3 g/L$   
 $M = ?$   
 $V = L \times w \times h = 25cm \times 8cm \times 10.4cm = 2080cm^3 = 2080mL = 2.08L$  (need units to agree)  
 $M = D \times V = (19.3 g/L) \times 2.08L = \boxed{40.1g}$

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$D = 1.85g/mL$   
 $M = 64g$   
 $V = ?$

$V = \frac{M}{D} = \frac{64g}{1.85g/mL} = \boxed{34.6mL}$

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Liquids are being combined, so the masses and volumes could be added up

$D = ?$   
 $M = 44.8g + 48.0g = 92.8g$   
 $V = 40mL + 50mL = 90mL$

$D = \frac{M}{V} = \frac{92.8g}{90mL} = \boxed{1.03g/mL}$

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	<u>Vol</u>	<u>Mass</u>	
A	48.5ml	37.2g	a) $D = \frac{M}{V}$
B	12.8ml	174.1g	
C	64.7ml	71.2g	

b) Liq. A will float to top (Lowest Density)  
 Liq B will sink to bottom (greatest Density)

(A)  $D = \frac{37.2g}{48.5ml} = .767g/mL$   
 (B)  $D = \frac{M}{V} = \frac{174.1g}{12.8ml} = 13.6g/mL$   
 (C)  $D = \frac{M}{V} = \frac{71.2g}{64.7ml} = 1.10g/mL$